This article was downloaded by:

On: 28 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713618290

Asymmetric Induction by Chiral Phosphorus

Kamyar Afarinkia^a; Hayley Binch^a; Ian Forristal^a; Clare Jones^a; James Lowman^a; Egizia De Pascale^a; Andrew Twist^a

^a Department of Chemistry, King's College, London, UK

Online publication date: 27 October 2010

To cite this Article Afarinkia, Kamyar , Binch, Hayley , Forristal, Ian , Jones, Clare , Lowman, James , De Pascale, Egizia and Twist, Andrew(2002) 'Asymmetric Induction by Chiral Phosphorus', Phosphorus, Sulfur, and Silicon and the Related Elements, 177: 6, 1641-1644

To link to this Article: DOI: 10.1080/10426500212218
URL: http://dx.doi.org/10.1080/10426500212218

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Phosphorus, Sulfur and Silicon, 2002, Vol. 177:1641–1644 Copyright © 2002 Taylor & Francis 1042-6507/02 \$12.00 + .00

DOI: 10.1080/10426500290093405



ASYMMETRIC INDUCTION BY CHIRAL PHOSPHORUS

Kamyar Afarinkia, Hayley Binch, Ian Forristal, Clare Jones, James Lowman, Egizia De Pascale, and Andrew Twist Department of Chemistry, King's College, The Strand, London, UK

(Accepted December 25, 2001)

We report on the asymmetric induction by a chiral phosphorus atom contained in a 2-oxo-1,3,2-oxazaphosphorinane ring into an adjacent keto, alkane, or alkene function.

Keywords: α -Ketophosphonate; asymmetric; oxazaphosphorinane; vinyl phosphonate

The objective of our research program is to develop asymmetric carbon-carbon bond-forming reactions of organophosphorus reagents containing a chiral phosphorus atom. We are studying the reduction and alkylation of asymetric α -ketophosphonates, such as 2-alkanoyl and 2-aroyl-2-oxo-1,3,2-oxazaphosphorinanes, because this functionality has a rich chemistry which can be further exploited to prepare a wide range of interesting structural motifs. We are also interested in the chemistry of asymmetric vinyl phosphonates. Nucleophilic addition to this function gives asymmetric β -substituted phosphonates, which can take part in a Horner-Wittig reaction to afford structural motifs found in many terpenoid natural products.

SYNTHESIS, CONFORMATIONS, AND REACTIONS OF 2-ALKANOYL AND 2-AROYL-2-OXO-1,3,2-OXAZAPHOSPHORINANES

2-Alkanoyl and 2-aroyl-2-oxo-1,3,2-oxazaphosphorinanes were prepared by an Arbuzov reaction of the corresponding 2-ethoxy-1,3,2-oxazaphosphorinanes and acid chlorides (Scheme 1).^{3–5}

We thank King's College, London, SmithKlineBeecham, Pfizer, The Royal Society, EPSRC (grant GR/L70066), GlaxoWellcome, and The Leverhulme Trust (grant F/40/AD) for financial support.

Address correspondence to Kamyar Afarinkia, Department of Chemistry, King's College, Strand, London WC2R 2LS, UK. E-mail: kamyar.afarinkia@kcl.ac.uk

SCHEME 1 Reagents and conditions: i, (EtO)PCl₂, Et₃N, CH₂Cl₂; ii, Aroylchloride or alkanoylchloride (excess).

The structure of a number of these α -ketophosphonates was studied by crystallography.⁵ The key structural features of these compounds is that, except when the nitrogen substituent is a trityl group, the ring adopts a pseudo-chair conformation. When the nitrogen substituent is a trityl group, the ring adopts a twisted boat conformation. In either case, the P=O bond occupies a psuedo-equatorial position^{5,6} and adopts a gauche or antirelationship with the C=O bond.

We expected that the chirality at the phosphorus atom in 2-alkanoyl-and 2-aroyl-2-oxo-1,3,2-oxazaphosphorinanes would result in an asymmetric reduction of the adjacent C=O function. This has proved to be true, although to date the levels of diastereoselectivity remain modest (about 50% de). Similarly, [3,3]-sigmatropic rearrangement of phosphoenol ethers afford asymmetric α -ketophosphonates with good diastereoselectivity (Scheme 2).

SCHEME 2 Reagents and conditions: i, KHMDS, THF, 0° C, R_2 CH=CHCH₂Br; ii, Toluene, reflux, 5 h.

ASYMMETRIC NUCLEOPHILIC ADDITION TO 2-PROPENYL-2-OXO-1,3,2-OXAZAPHOSPHORINANES

We have previously reported on the nucleophilic additions to chiral vinylphosphonates using organocuprates, with chlorotrimethylsilane and tetramethylethylenediamine as rate-accelerating additives.^{7,8} The diastereomeric purity of the products, after chromatography and a single crystallization from petroleum ether, were excellent (Scheme 3).^{8a}

We have also investigated addition to enantiopure vinyl phosphonates derived from N-trityl-(1R,2S)-norephedrine. The S_P isomer was

$$\begin{array}{c} & & & & \\ & & & \\ & & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & &$$

SCHEME 3 Reagents and conditions: i, RMgBr, CuI, TMSCl, TMEDA, -78° C to -10° C; ii, TBAF, THF, r.t.

isolated as a single enantiomer. However, the diastereoselectivity of the additions to this five-membered ring analogue proved significantly less than the six-membered ring analogue.⁹

ALKYLATION OF 2-OXO-1,3,2-OXAZAPHOSPHORINANES

As described before, we intend to use the asymmetric β -substituted phosphonates prepared in the previous section in Horner-Wittig reactions. As a preamble to this study, we were interested in investigating the diastereoselectivity of the reaction of these phosphorus stablized anions with electrophiles.

We prepared a series of 2-alkyl-2-oxo-1,3,2-oxazaphosphorinanes with and without a substituent at their β position and investigated the diastereoselectivity in the alkylation at the α position.

Alkylation of 2-alkyl-2-oxo-1,3,2-oxazaphosphorinanes without a substituent at their β position proceeds with good selectivity. This selectivity is influenced by the steric bulk of the nitrogen substituent as expected. Stereochemistry of the β -substituent either enhances (in the case of R,R/S,S pair) or counteracts (in the case of R,S/S,R pair) the stereoselectivity arising from the asymmetric phosphorus atom itself. Cooperative effects from the chirality of the phosphorus atom and its β substituent results in excellent degrees of stereoselectivity in these reactions (Scheme 4).

E = Allyl (35%, d.r. > 15); E = Me (58%, d.r. > 4); E = Bn (38%, d.r. > 7); E = Me Si (44%, d.r. = 1)

SCHEME 4 Reagents and conditions: LDA, toluene, E-Br -78° C to 0° C.

ACKNOWLEDGEMENTS

We thank Prof. David Williams and Dr. Andrew White (Imperial College, London), and Dr. Jon. W. Steed (King's College, London), for X-ray crystallography.

REFERENCES

- K. Afarinkia and M. V. Vinader, Organic Functional Group Transformations, C. J. Moody (ed.) (Pergamon, London, 1995), Vol. 5, p. 393.
- [2] B. Bartels, J. Clayden, C. Gonzalez-Martin, A. Nelson, M. G. Russell, and S. Warren, J. Chem. Soc. Perkin. Trans. I, 1807 (1999), and references cited therein.
- (a) S. E. Denmark and C.-T. Chen, J. Org. Chem., 59, 2922 (1994); (b) C. J. Cramer,
 S. E. Denmark, P. C. Miller, and R. L. Darrow, J. Am. Chem. Soc., 116, 2437 (1994),
 and references cited therein.
- [4] N. J. Gordon and S. A. Evans, Jr., J. Org. Chem., 58, 5293 (1993).
- [5] K. Afarinkia, R. Angell, C. L. Jones, and J. Lowman, Tetrahedron Lett., 42, 743 (2001).
- [6] Y. Huang, A. M. Arif, and W. G. Bentrude, J. Org. Chem., 58, 6235 (1993), and references cited therein.
- [7] K. Afarinkia, H. M. Binch, and C. Modi, Tetrahedron Lett., 39, 7419 (1998).
- [8] (a) K. Afarinkia, H. M. Binch, C. L. Jones, and E. De Pascale, Synlett, 1769 (2000);
 (b) K. Afarinkia, H. M. Binch, C. L. Jones, and E. De Pascale, J. Chem. Soc. Perkin. Trans. 1, manuscript submitted.
- [9] K. Afarinkia, H. M. Binch, and I. Forristal, Synlett, 1771 (2000).